

## IN THE SPECIFICATION

*Kindly replace Paragraphs [0001] to [0009] with the following:*

### Technical Field

[0001] The present invention relates to ferritic Cr-contained steel having a low thermal expansion coefficient[,] and particularly The disclosure also relates to ferritic Cr-contained steel having a low thermal expansion coefficient suitable for applications in which a heat cycle is repeated between high temperature and low temperature, including exhaust system members of an automobile such as exhaust manifolds, exhaust pipes, converter case materials, and metal honeycomb materials; separators within a solid-oxide-type fuel cell; materials for interconnectors; materials for reformers as peripheral members of fuel cells; exhaust ducts of power generation plants; or heat exchangers. ~~All the~~ The thermal expansion coefficients ~~described in the invention is the~~ herein are linear expansion coefficients. It will hereinafter be abbreviated as thermal expansion coefficient.

### Background Art

[0002] In various members subjected to the repeated heat cycle between high temperature and low temperature, heat expansion and contraction are repeated, as a result both of the members themselves and peripheral members of them are added with strain or stress, and consequently fracture by thermal fatigue is prone to occur. In such a circumstance, the fracture by thermal fatigue is hardly to occur in an alloy having a lower thermal expansion coefficient, because heat strain and heat stress to be added become smaller. As a known method for decreasing the thermal expansion coefficient, use of Magneto-volume effects is given. This is a method for decreasing the thermal expansion coefficient in such a way that when temperature is decreased, strain corresponding to a level of essentially contracted strain is compensated by magnetostriction due to generation of Atomic magnetic momentum or change in amount of the momentum. To obtain such magneto-volume effects, temperature dependence of the generation

or the change in amount of the atomic magnetic-momentum is important. For example, in Fe-36% Ni Invar alloy used for a shadow mask in a cathode ray tube of a display, since the amount of the Atomic magnetic momentum suddenly changes near the Curie temperature (230 to 279°C), a sudden decrease in thermal expansion coefficient is exhibited at a temperature lower than the Curie temperature (a value of thermal expansion coefficient of the alloy at about 200°C, at which the alloy is used for the shadow mask, is extremely low, about  $1 \times 10^{-6}$  /°C.) However, the alloy has an extremely high thermal expansion coefficient of about  $18 \times 10^{-6}$  /°C at 800°C, which is in at the same level as in a typical austenitic stainless steel. Furthermore, the alloy contains Ni as much as 36%, resulting in an extreme increase in cost, consequently it is hard to be used for such an application in general consumer goods. From such reasons, Fe-Cr base alloys are widely used for the application. However, the Fe-Cr base alloys have a small temperature dependence of amount of the Atomic magnetic momentum is small, therefore the Magneto-volume effect is not observed even at a temperature of the Curie temperature or lower. In this way, decrease in thermal expansion coefficient due to Magneto-volume effect is difficult in the Fe-Cr base alloys. Therefore, in the related art, thermal fatigue life has been improved by a method using improvement in strength or high ductility by forming a high alloy (JP-A-2003-213377 and JP-A-2002-212685). However, improved strength by forming the high alloy necessarily causes a problem of reduction in workability, and orientation of high ductility causes strength to be extremely lowered, consequently it is pointed that another problem (for example, fatigue at elevated temperature) may occur. From such a situation, a new method has been strongly required for improving the thermal fatigue life by reducing the thermal expansion coefficient of Fe-Cr ferritic alloys.

#### Disclosure of the Invention Summary

[0003] ~~An object of the invention is to decrease the thermal expansion coefficient of the Fe-Cr ferritic alloys.~~

[0004] ~~The inventors have earnestly carried out the research to achieve the object, as a result~~ We found that addition of W to the Fe-Cr ferritic alloys and a decrease in the amount of

precipitated W remarkably contributed to a decrease in thermal expansion coefficient of the alloys. While a the mechanism of this has not been clarified, since it is known that the thermal expansion coefficient of the alloys also depends on specific heat and bulk modulus, it is considered believed that addition of W has an effect on the coefficient through the temperature dependence of the amount of the Atomic magnetic momentum. Especially An especially important point is that simple addition of W is not sufficient, and large amount of precipitated W rather increases the thermal expansion coefficient. The precipitated state of W is a precipitated state mainly in a form of the Laves phase (Fe2M-type intermetallic compounds) or carbides, and when W is in a state of precipitated W, it inhibits a decrease in the thermal expansion coefficient. While the reason for this is not clear, the inventors estimate we believe it is because of the following two points. The first point is considered as follows: while grain boundaries essentially act as a cushion for thermal expansion, since the Laves phase is precipitated therein, the cushion effect is reduced, and consequently the thermal expansion coefficient is increased.

The second point is considered as follows: when the amount of the precipitated W is increased in the alloy, the amount of solid soluted W is decreased, and consequently a decrease in the thermal expansion coefficient of the alloy is inhibited. However, even if the amount of precipitated W is slight, for example, only more than 0.1%, the decrease in thermal expansion coefficient of the alloy is inhibited, therefore the reason can not be explained only from the increase in the amount of dissolved W in the alloy. Thus, the former reason, a decrease in effect as a cushion of the grain boundaries is considered to be major. However, detailed study will be still necessary on these reasons. In this way, the knowledge of decrease in thermal expansion coefficient by controlling a state of W was obtained. Therefore, component design of a material suitable for the environment in which heat cycle is applied can be realized by considering the knowledge on thermal expansion coefficient in addition to knowledge in the related art, that is, influence of various additional-elements on other properties such as workability, oxidation resistance, and corrosion resistance.

[0005] The invention was made based on the knowledge, and is summarized as follows.  
Select aspects of the disclosure include:

1. Ferritic Cr-contained steel containing C of 0.03% or less, Mn of 5.0% or less, Cr of 6 to 40%, N of 0.03% or less, Si of 5% or less, and W of 2.0% to 6.0% in percent by mass, and Fe and inevitable impurities as the remainder, wherein precipitated W is 0.1% or less in percent by mass, and an average thermal expansion coefficient between 20°C and 800°C is less than  $12.6 \times 10^{-6} /^{\circ}\text{C}$ .
2. The ferritic Cr-contained steel according to 1, further containing at least one selected from a group of Nb of 1% or less, Ti of 1% or less, Zr of 1% or less, Al of 1% or less, and V of 1% or less in percent by mass.
3. The ferritic Cr-contained steel according to 1 or 2, further containing Mo of 5.0% or less in percent by mass.
4. The ferritic Cr-contained steel according to any one of 1 to 3, further containing at least one selected from a group of Ni of 2.0% or less, Cu of 3.0% or less, and Co of 1.0% or less in percent by mass.
5. The ferritic Cr-contained steel according to any one of 1 to 4, further containing at least one selected from a group of B of 0.01% or less and Mg of 0.01% or less in percent by mass.
6. The ferritic Cr-contained steel according to any one of 1 to 5, further containing one or two of REM of 0.1% or less and Ca of 0.1% or less in percent by mass.
7. A manufacturing method of ferritic Cr-contained steel, wherein a composition of molten steel is adjusted to include C of 0.03% or less, Mn of 5.0% or less, Cr of 6 to 40%, N of 0.03% or less, Si of 5% or less, and W of 2.0% to 6.0% in percent by mass, and Fe and inevitable impurities as the remainder; and then the molten steel is formed into a steel slab; and then the slab is hot-rolled and then subjected to hot-rolled-sheet annealing at a hot-rolled-sheet annealing temperature of 950 to 1150°C and descaling; and furthermore, a hot rolled and annealed sheet is cold-rolled and then subjected to finish annealing at a finish annealing temperature of 1020°C to 1200°C, so that precipitated W is 0.1% or less in percent by mass.
8. The manufacturing method of ferritic Cr-contained steel according to 7, wherein the composition of the molten steel further includes at least one selected from a group of Nb of 1% or less, Ti of 1% or less, Zr of 1% or less, Al of 1% or less, and V of 1% or less in percent by mass.

9. The manufacturing method of ferritic Cr-contained steel according to 7 or 8, wherein the composition of the molten steel further includes Mo of 5.0% or less in percent by mass.
10. The manufacturing method of ferritic Cr-contained steel according to 7 to 9, wherein the composition of the molten steel further includes at least one selected from a group of Ni of 2.0% or less, Cu of 3.0% or less, and Co of 1.0% or less in percent by mass.
11. The manufacturing method of ferritic Cr-contained steel according to 7 to 10, wherein the composition of the molten steel further includes at least one selected from a group of B of 0.01% or less and Mg of 0.01% or less in percent by mass.
12. The manufacturing method of ferritic Cr-contained steel according to 7 to 11, wherein the composition of the molten steel further includes one or two of REM of 0.1% or less and Ca of 0.1% or less in percent by mass.

[0006] While the amount of “precipitated W” ~~in the invention~~ means mass percent of W precipitated mainly in a form of the Laves phase or carbides, mass percent of W precipitated in a form of another phase is also included. The mass percent of “precipitated W” was measured by inductively coupled plasma atomic emission spectrometry (ICP-AES). That is, a sample is electrolyzed at a constant current (current density $\leq$ 20 mA/cm<sup>2</sup>) using a 10% acetylacetone-base electrolyte (commonly called AA solution). Electrolysis residue in the electrolytic solution is collected by filtration, then fused in alkali (sodium peroxide and metaboric lithium), and then dissolved in an acid and then diluted into a certain quantity by water. The solution is subjected to measurement of the amount of W (W<sub>p</sub>) in the solution using an ICP emission spectrometer (Inductively Coupled Plasma Spectrometer). The amount of precipitated W (mass percent) can be obtained by the following formula;

$$\text{the amount of precipitated W (mass percent)} = (\text{W}_p / \text{sample weight}) \times 100.$$

[0007] The thermal expansion coefficient has temperature dependence even if a ferrite structure is remained as it is. Thus, average thermal expansion coefficient in ~~use a real world~~ environment is practically important. Therefore, ~~the invention~~-we defined an average thermal expansion coefficient between 20°C and 800°C. The average thermal expansion coefficient

between 20°C and 800°C described herein means a value of an elongation ratio in one direction of a steel sheet in the case of heating the steel sheet to 20°C to 800°C which is divided by temperature difference 780°C between 20°C and 800°C. However, since the ~~invention Cr-contained steels~~ effectively acts on decrease in thermal expansion coefficient even out of the temperature range, it will be appreciated that the limitation of the temperature range is not intended to limit the temperature in ~~use~~ a real world environment to the range of 20°C to 800°C.

[0008] According to the ~~invention~~, ferritic Ferritic Cr-contained steel having a low thermal expansion coefficient compared with ferritic Cr-contained steel in the related art can be obtained. Thermal fatigue life at 100 to 800°C of such a material having a low thermal expansion coefficient exhibits an excellent value compared with steels in the related art (ferritic stainless steel, Type 429Nb (JIS G4307) and ferritic heat-resistant steel, sheet SUH409L (JIS G4312)).

[0009] Therefore, the steel ~~of the invention~~ is used in a region to which heat cycle is applied, thereby thermal stress to the peripheral member and the steel itself is reduced, and therefore a problem in design for improving the life, or complicated design for reducing the thermal strain is not necessary. Therefore, the ~~invention steels~~ can be preferably used for applications of components to which heat cycle is applied, including the exhaust system components of the automobile, separators within the fuel cell, materials for interconnectors, materials for reformers, exhaust ducts of the power generation plants, or heat exchangers.

*Kindly replace Paragraphs [0015] to [0031] with the following:*

Best Mode for Carrying Out the Invention Detailed Description

[0015] Hereinafter, the reason for ~~limiting selecting elements to be in~~ the composition to be within the above range ~~in the invention~~ is described. In the description, representation in “%” is representation in mass percent unless otherwise specified.

- C: 0.03% or less

[0016] Since C deteriorates toughness and workability, incorporation of C is preferably reduced at maximum. From the point, the amount of C was limited to about 0.03% or less in the invention. Preferably, the amount is about 0.008% or less.

- Mn: 5.0% or less

[0017] Mn is added for improving toughness. To obtain the effect, the amount of Mn of 0.1% or more is preferable. However, since excessive addition of Mn may cause formation of MnS, which deteriorates corrosion resistance, the amount was limited to about 5.0%, or less. Preferably the amount is about 0.1% to 5.0%, and more preferably about 0.5% to about 1.5%.

- Cr: 6 to 40%

[0018] Cr is also effective for improving corrosion resistance and oxidation resistance. In the invention, since Since W of 2.0% or more is added, if Cr of 6% or more exists in steel, the steel can be used for many applications from a point of corrosion resistance or oxidation resistance. In particular, when high-temperature oxidation resistance is regarded as important, Cr of 14% or more is preferably contained. When the amount of Cr exceeds 40%, embrittlement in material becomes significant; therefore the amount was determined to be about 40% or less. When workability is regarded as important, the amount of Cr is preferably less than about 20%, and more preferably less than about 17%.

[0019] Moreover, Cr is effective for decrease in thermal expansion coefficient, and in the light of this point, the amount of about 14% or more is preferable.

- N: 0.03% or less

[0020] Since N deteriorates toughness and workability similarly as C, incorporation of N is preferably reduced at maximum. From this point, the amount of N was limited to about 0.03% or less ~~in the invention~~. More preferably, the amount is about 0.008% or less.

- Si: 5% or less

[0021] Si is added for improving oxidation resistance. To obtain the effect, the amount of Si is preferably 0.05% or more. When the amount exceeds 5%, strength at room temperature is increased, which deteriorates workability, therefore the upper limit of the amount was determined to be about 5%. Preferably, the amount is about 0.05% to about 2.00%.

- W: 2.0% to 6.0%

[0020] W is an extremely important element ~~in the invention~~. Since addition of W largely reduces thermal expansion coefficient, the amount of W was determined to be about 2.0% or more. However, when the amount is excessively increased, strength at room temperature is increased, which deteriorates workability, therefore the upper limit of the amount was determined to be about 6.0%. Preferably, the amount is about 2.5% to about 4%, and more preferably about 3% to about 4%.

- Precipitated W: 0.1% or less

[0023] The precipitated W is precipitated mainly in the form of the Laves phase or carbides. When the precipitated W exceeds 0.1%, the effect of decrease in thermal expansion coefficient due to addition of W is small. Therefore, the upper limit of the amount of precipitated W was determined to be about 0.1% or less. Preferably, the amount is about 0.05% or less, and more preferably about 0.03% or less. The lower amount is more preferable. However, finish annealing temperature must be increased significantly in order to restrain the precipitated W to be less than 0.005%, which results in extremely coarsened crystal grains, consequently orange peel occurs during working, causing a crack during working. Therefore,

particularly when the steel of the invention of the application is used for an application requiring working, it is more preferable that the amount of precipitated W is substantially about 0.005% or more. While the amount of "precipitated W" means mass percent of W precipitated mainly in the form of the Larves phase or carbides, it may include mass percent of W precipitated in a form of another phase. In measurement of the mass percent of "precipitated W", the electrolysis residue was measured in the inductively coupled plasma atomic emission spectrometry as described before.

[0024] Hereinbefore, while basic components have been described, in addition to this, the following elements can be appropriately contained as necessary in the invention.

- At least one selected from Nb of about 1% or less, Ti of about 1% or less, Zr of about 1% or less, Al or about 1% or less, and V of about 1% or less

[0025] Any of Nb, Ti, Zr, Al and V acts to fix C or N and thus improves intergranular corrosion resistance, and from this point, each of them is preferably contained about 0.02% or more. However, when the amount exceeds 1%, embrittlement of steel is caused; therefore they are determined to be contained about 1% or less respectively.

- Mo: 5.0% or less

[0026] Mo may be added because it improves corrosion resistance. While the effect appears at the amount of about 0.02% or more, excessive addition of Mo deteriorates workability, therefore the amount of about 5.0% was determined as the upper limit. The amount is preferably about 1% to about 2.5%.

- At least one selected from Ni of 2.0% or less, Cu of 3.0% or less, and Co of 1.0% or less

[0027] Any of Ni, Cu, and Co is a useful element for improving toughness, and Ni of about 2.0% or less, Cu of about 3.0% or less, and Co of about 1.0% or less were determined to

be contained respectively. Ni of about 0.5% or more, Cu of about 0.3% or more, and Co of about 0.01% or more are preferably added so that effects of the elements are sufficiently exhibited.

- At least one selected from B of 0.01% or less and Mg of 0.01% or less

[0028] Both of B and Mg effectively contribute to improvement in secondary embrittlement. To obtain the effect, B of about 0.0003% or more and Mg of about 0.0003% or more are preferable respectively. However, in each of B and Mg, when the amount exceeds about 0.01%, strength at room temperature is increased, causing deterioration in ductility, therefore they are determined to be contained about 0.01% or less respectively. More preferably, B is about 0.002% or less, and Mg is about 0.002% or less.

- At least one of REM of 0.1% or less and Ca of 0.1% or less

[0029] REM and Ca effectively contribute to improvement in oxidation resistance. To obtain the effect, REM of about 0.002% or more and Ca of about 0.002% or more are preferable respectively. However, since excessive addition of them deteriorates corrosion resistance, they are determined to be contained about 0.1% or less respectively. In the invention, REM means lanthanoid series elements and Y. In particular, when Ti is contained, Ca effectively contributes also to prevention of nozzle clogging during continuous casting. The effect becomes significant at the Ca amount of about 0.001% or more.

[0030] Next, a microstructure of a steel sheet is described. A structure of steel manufactured using a technique of the application is substantially a ferrite single phase. While the steel may have a structure partially containing bainite, in a condition that cooling has been performed after hot rolling and coiling, steel after cold rolling and annealing substantially has the structure of the ferrite single phase. In the steel ~~of the invention~~ of the application, component design is made such that hard martensite is not formed in a condition before working such as cold rolling and annealing.

[0031] Next, a preferred manufacturing method of the steel ~~of the invention~~ is described. Manufacturing conditions of the steel ~~of the invention~~ is not particularly limited except that the hot-rolled sheet annealing temperature and the finish annealing temperature are determined to obtain precipitated W of 0.1% or less, and a typical manufacturing method of the ferritic stainless steel can be preferably used.

*Kindly replace paragraphs [0033] and [0034] with the following:*

[0033] In a more preferable manufacturing method, part of conditions of a hot rolling process and a cold rolling process are made to be specific conditions. In steel making, it is preferable that molten steel containing the essential components and components added as necessary is ingot in the converter or the electric furnace, and then an ingot is subjected to secondary refining by a VOD method. While the molten steel formed into the ingot can be formed into a steel material according to a known manufacturing method, continuous casting is preferably used in the light of productivity and quality. A steel material obtained by the continuous casting is heated, for example, to about 1000 to about 1250°C, and then formed into a hot-rolled sheet having a desired thickness. Naturally, the material can be worked into other forms than a sheet material. The hot-rolled sheet is subjected to batch annealing or continuous annealing at about 950 to about 1150°C, and more preferably about 1020 to about 1150°C, and then descaled by pickling and the like to be formed into a hot-rolled sheet product. Shot blasting may be performed for descaling before pickling as necessary.

[0034] Furthermore, the obtained hot rolled and annealed sheet is formed into a cold-rolled sheet through the cold rolling process. In the cold rolling process, at least two steps of cold rolling including intermediate annealing may be performed as necessary for production reasons. Total reduction rate during the cold rolling process including one or at least two steps of cold rolling is made to be about 60% or more, preferably about 62% or more, and more preferably about 70% or more. A cold rolled sheet is subjected to continuous annealing (finish annealing) at about 1020°C to about 1200°C and more preferably about 1050°C to about 1150°C, and then subjected to pickling to be formed into a cold rolled and annealed sheet. In

some applications, light rolling (for example, skin-pass rolling) can be applied after cold rolling and annealing to adjust a shape of the steel sheet or quality.

*Kindly replace paragraph [0036] with the following.*

[0036] Particularly, ~~in the invention~~, it is important to determine the hot-rolled sheet annealing temperature and the finish annealing temperature to obtain precipitate W of 0.1% or less.

(1) Hot-rolled-sheet annealing temperature: 950°C to 1150°C, and finish annealing temperature: 1020°C to 1200°C.

*Kindly replace paragraph [0038] with the following.*

Embodiment-1 Example 1

[0038] 50 kg steel ingots having compositions as shown in Table 1 (examples according to selected aspects of the invention, comparative steels and steels in the related art (Type 429Nb, SUH409L)) were prepared, and then these steel ingots were heated to 1100°C, and then formed into hot rolled sheets 4 mm in thickness by hot rolling. Next, the hot rolled sheets were sequentially subjected to hot-rolled-sheet annealing (annealing temperature: 1090°C), pickling, cold rolling (reduction rate: 62.5%), finish annealing (annealing temperature was changed from 900°C to 1220°C as shown in Table 1, and the sheets were held for three minuets at respective temperatures, and then air-cooled, so that the amount of precipitated W was adjusted), and pickling, consequently 1.5 mm thick steel sheets were formed.

*Kindly replace paragraphs [0047] and [0048] with the following.*

Embodiment-2 Example 2

[0047] Next, a relation between the amount of precipitated W and the hot-rolled-sheet annealing temperature was investigated. A 50 kg steel ingots having a composition of C of 0.005%, Si of 0.07%, Mn of 1.02%, Cr of 15.2%, Mo of 1.92%, W of 3.02%, Nb of 0.51% and N of 0.004% were prepared, and then these steel ingots were heated to 1100°C, and then formed into hot rolled sheets 4 mm in thickness. Next, the hot rolled sheets were sequentially subjected to hot-rolled-sheet annealing (annealing temperature was changed from 900°C to 1200°C, and the sheets were held for three minuets at respective temperatures, and then air-cooled), pickling, cold rolling (reduction rate: 62.5%), finish annealing (the sheets were held for three minuets at the finish annealing temperature of 1100°C, and then air-cooled), and pickling, consequently 1.5 mm thick steel sheets were formed.

[0048] The amount of precipitated W in the cold rolled and annealed sheets obtained in this way were measured in the same manner as in the ~~embodiment~~ Example 1. Test pieces for evaluation of the amount of precipitated W were sampled from two points in respective steel sheets, and each average value of the two was determined as a value of precipitated W.

*Please delete paragraph [0050].*